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Relation Between Composition and Density of Aqueous Solutions of Copper Sulphate and Sulphuric Acid Homer Denison Holler 1917

Sulfur Dioxide and Its Aqueous Solutions Carol E. Maass 1928

Schedule for Density-composition Tables for Aqueous Solutions of Nitric Acid British Standards Institution 1987

Density-composition Tables for Aqueous

Solutions of Hydrochloric Acid British Standards Institution 1957

On the Variation of the Density with the Concentration of Weak Aqueous Solutions of Certain Salts James Gordon MacGregor 1889

Electrochemistry of High Density Light Metals in Aqueous Solutions Unesco

Viscosity and Density of Aqueous Solutions of LiBr, LiCl, ZnBr2, CaCl2 and LiNo3 Martin Wimby 1993

Schedule for Density-Composition Tables for Aqueous Solutions of Sulphuric Acid British Standards Institute Staff 1987-12-23 Sulfuric acid, Solutions, Density, Hydrometers, Surface tension, Temperature, Error correction, Errors, Mass, Concentration (chemical)

British Standard Density-composition Tables for Aqueous Solutions of Nitric Acid British Standards Institution 1957

Densities and Refractive Indices of Aqueous Solutions. Diethylenetriamine, Triethylenetetramine, and Tetraethylenepentamine Denis J. Rouleau 1962 Continuing an investigation of the effect of chemical structure on density and refractive index, data were obtained for aqueous solutions of diethylenetriamine, triethylenetetramine, and tetraethylenepentamine. The purification procedures and methods of analysis described by Chu and Thompson were employed. Refractive index and density data were determined at 25C. (Author).

Schedule for Density-Composition Tables for Aqueous Solutions of Nitric Acid British

Standards Institute Staff 1987-12-23 Nitric acid, Solutions, Hydrometers, Density, Errors, Error correction, Surface tension, Temperature, Mass, Concentration (chemical)

On the Density Measurements of Aqueous Solutions of Potassium Palmitate Glenn Edwin Welde 1941

Density-composition Tables for Aqueous Solutions of Sul[British Standards Institution 1987

Density-composition Tables for Aqueous Solutions of Nitric Acid British Standards Institution 1957

On the Density of Aqueous Solutions of Lauryl Sulphonic Acid Harold Eugene Kirkby 1943

Study of the Partial Density of the Solvent in Aqueous Solutions of Electrolytes Lengyel, S 1961

Density-composition Tables for Aqueous Solutions of Sulphuric Acid British Standards Institution 1987

Study of the Partial Density of the Solvent in Aqueous Solution of Electrolytes S. Lengyel 1961

On the Density of Aqueous Solutions of Potassium Palmitate Herman J. Byrd 1940

Viscosity and Density of Aqueous Solutions of LiBr, LiCl, ZnBr2, CaCl2 and LiNo3 Martin Wimby 1993

Densities of Aqueous Solutions of Inorganic Substances Otakar Söhnel 1985

On the Density of Weak Aqueous Solutions of Certain Sulphates James Gordon MacGregor 1890

Density Prediction of Multicomponent Aqueous Solutions from Binary Data 1975

Density-composition Tables for Aqueous Solutions of Sulphuric

Acid for Use in Conjunction with British Standard Density Hydrometers British Standards Institution 1937

A Study of Human Low Density Lipoproteins in Aqueous Solutions Using Dielectric Methods G. S. Chana 1980

Viscosity and Density of Aqueous Solutions of LiBr, LiCl, ZnBr2, CaCl2 and LiNo3 Martin Wimby 1993

Density-composition Tables for Aqueous Solutions of Sodium Chloride and of Calcium Chloride British Standards Institution 1938

Density of Aqueous Solutions of CO2 2001 In this report, we present a numerical representation for the partial molar volume of CO2 in water and the calculation of the corresponding aqueous solution density. The motivation behind this work is related to the importance of having accurate representations for aqueous phase properties in the numerical simulation of carbon dioxide disposal into aquifers as well as in geothermal applications. According to reported experimental data the density of aqueous solutions of CO2 can be as much as 2-3% higher than pure water density. This density variation might produce an influence on the groundwater flow regime. For instance, in geologic sequestration of CO2, convective transport mixing might occur when, several years after injection of carbon dioxide has stopped, the CO2-rich gas phase is concentrated at the top of the formation, just below an overlaying caprock. In this particular case the heavier CO2 saturated water will flow downward and will be replaced by water with a lesser CO2 content.

Density (specific Gravity and Thermal Expansion Under Atmospheric Pressure) of Aqueous Solutions of Inorganic Substances and of Strong Electrolytes, by J.A. Beattie, L.J. Gillespie, G. Scatchard, W.C. Schumb and R.F. Tefft Ralph F. Tefft 1928

On the Density of Weak Aqueous Solutions of Nickel Sulphate James Gordon MacGregor 1891

Density-composition Tables for Aqueous Solutions of Sodium Hydroxide British Standards Institution 1961

Density-composition Tables for Aqueous Solutions of Caustic Soda for Use in Conjunction with British Standard Density Hydrometers British Standards Institution 1938

A Thermodynamic Model for Aqueous Solutions of Liquid-like Density 1987 The paper describes a model for the prediction of the thermodynamic properties of multicomponent aqueous solutions and discusses its applications. The model was initially developed for solutions near room temperature, but has been found to be applicable to aqueous systems up to 300°C or slightly higher. A liquid-like density and relatively small compressibility are assumed. A typical application is the prediction of the equilibrium between an aqueous phase (brine) and one or more solid phases (minerals). (ACR).

Schedule for Density-Composition Tables for Aqueous Solutions of Hydrochloric Acid British Standards

Institute Staff 1987-12-31 Density, Hydrochloric acid, Errors, Error correction, Temperature, Surface tension, Solutions, Hydrometers, Mass, Concentration (chemical)

Viscosity and Density of Aqueous Solutions of Urea and Guanidine Hydrochloride Kazuo Kawahara 1966

The Accurate Determination of Densities of Aqueous Solutions David R. Schink 1958

A Study of the Density of Aqueous Solutions of Nickel Sulphate Hueston Marion Louderback 1940

Schedule for Density-composition Tables for Aqueous Solutions of Hydrochloric Acid British Standards Institution 1987

British Standard Density-composition Tables for Aqueous Solutions of Sulphuric Acid British Standards Institution 1959

Thermophysical Properties of Lithium Bromide + 1, 2-Propanediol Aqueous Solutions Solubility, Density and Viscosity 2012 The solubilities, densities and viscosities of lithium bromide (LiBr) + 1, 2-propanediol (HO-CH2-CHOH-CH3) aqueous solution (mass ratio of LiBr/HO-CH2-CHOH-CH3 = 3.5, 4.5 and 5.5) were measured in the mass fraction range from 0.30 to 0.75. Solubility measurements were performed by the visual method in the temperature range of (271.15 to 345.15) K. The density measurements were made using an automated vibrating tube density meter, and the viscosity measurements were carried out with an automated falling-ball viscometer in the temperature range of (293.15 to 363.15) K.

The density and viscosity data were correlated with appropriate regression equations as a function of the mass fraction and temperature. The maximum average absolute deviations (AAD) between experimental and correlated data were 0.08% and 1.51% for densities and viscosities, respectively.